

$S = 1.256$
1295 reflections
217 parameters
H-atom parameters not
refined
 $w = 1/[\sigma^2(F_o) +$
 $0.00191(F_o)^2]$

Extinction correction:
Larson (1967)
Extinction coefficient:
0.012 (2)
Atomic scattering factors
from *SHELXTL-Plus*
(Sheldrick, 1991)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)

$$U_{eq} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* a_i \cdot a_j$$

	x	y	z	U_{eq}
C(1)	1.0614 (4)	0.4048 (2)	-0.0659 (1)	0.050 (1)
C(2)	1.0640 (4)	0.4626 (2)	-0.1247 (1)	0.052 (1)
C(3)	0.8671 (3)	0.4995 (2)	-0.1423 (1)	0.041 (1)
C(4)	0.7758 (3)	0.5823 (2)	-0.0994 (1)	0.040 (1)
C(5)	0.7853 (3)	0.5291 (2)	-0.0381 (1)	0.036 (1)
C(6)	0.7056 (4)	0.6047 (2)	0.0095 (1)	0.046 (1)
C(7)	0.7008 (4)	0.5458 (2)	0.0662 (1)	0.038 (1)
C(8)	0.8007 (3)	0.4532 (2)	0.0780 (1)	0.049 (1)
C(9)	0.9412 (3)	0.4036 (2)	0.0359 (1)	0.039 (1)
C(10)	0.9797 (3)	0.4791 (2)	-0.0178 (1)	0.036 (1)
C(11)	1.1255 (4)	0.3681 (3)	0.0671 (1)	0.052 (1)
C(12)	1.0897 (4)	0.2992 (2)	0.1211 (1)	0.052 (1)
C(13)	0.9590 (4)	0.3569 (2)	0.1650 (1)	0.039 (1)
C(14)	0.7727 (4)	0.3884 (2)	0.1335 (1)	0.041 (1)
C(15)	1.0445 (4)	0.4677 (2)	0.1887 (1)	0.039 (1)
C(16)	1.2524 (4)	0.4689 (2)	0.2057 (1)	0.053 (1)
C(17)	0.9031 (4)	0.2737 (2)	0.2125 (1)	0.050 (1)
C(18)	0.5642 (4)	0.5955 (3)	-0.1157 (1)	0.063 (1)
C(19)	0.8701 (5)	0.6993 (2)	-0.1052 (1)	0.057 (1)
C(20)	1.1221 (4)	0.5730 (2)	-0.0022 (1)	0.055 (1)
O(1)	0.8805 (3)	0.5465 (2)	-0.1992 (1)	0.053 (1)
O(2)	0.9276 (3)	0.5053 (1)	0.2351 (1)	0.056 (1)
O(3)	1.2931 (4)	0.3898 (2)	0.2496 (1)	0.087 (1)
O(4)	1.0588 (4)	0.2225 (2)	0.2424 (1)	0.073 (1)

Table 2. Selected geometric parameters (\AA , $^\circ$)

C(1)—C(2)	1.530 (3)	C(2)—C(3)	1.502 (3)
C(3)—C(4)	1.539 (3)	C(4)—C(5)	1.557 (3)
C(5)—C(6)	1.530 (3)	C(6)—C(7)	1.491 (3)
C(7)—C(8)	1.332 (3)	C(8)—C(9)	1.507 (3)
C(1)—C(10)	1.533 (3)	C(5)—C(10)	1.557 (3)
C(9)—C(10)	1.559 (3)	C(9)—C(11)	1.538 (4)
C(11)—C(12)	1.518 (4)	C(12)—C(13)	1.530 (3)
C(8)—C(14)	1.513 (3)	C(13)—C(14)	1.541 (3)
C(13)—C(15)	1.548 (3)	C(15)—C(16)	1.508 (4)
C(13)—C(17)	1.532 (3)	C(4)—C(18)	1.536 (4)
C(4)—C(19)	1.545 (4)	C(10)—C(20)	1.539 (4)
C(2)—C(1)—C(10)	113.2 (2)	C(1)—C(2)—C(3)	111.3 (2)
C(2)—C(3)—C(4)	113.1 (2)	C(2)—C(3)—O(1)	107.7 (2)
C(4)—C(3)—O(1)	111.9 (2)	C(3)—C(4)—C(5)	108.3 (2)
C(3)—C(4)—C(18)	107.8 (2)	C(5)—C(4)—C(18)	107.9 (2)
C(3)—C(4)—C(19)	110.0 (2)	C(5)—C(4)—C(19)	115.4 (2)
C(18)—C(4)—C(19)	107.3 (2)	C(4)—C(5)—C(6)	113.9 (2)
C(4)—C(5)—C(10)	117.9 (2)	C(6)—C(5)—C(10)	108.9 (2)
C(5)—C(6)—C(7)	111.7 (2)	C(6)—C(7)—C(8)	123.8 (2)
C(7)—C(8)—C(9)	122.2 (2)	C(7)—C(8)—C(14)	121.8 (2)
C(9)—C(8)—C(14)	115.9 (2)	C(8)—C(9)—C(10)	113.9 (2)
C(8)—C(9)—C(11)	110.4 (2)	C(10)—C(9)—C(11)	113.0 (2)
C(1)—C(10)—C(5)	109.0 (2)	C(1)—C(10)—C(9)	108.2 (2)
C(5)—C(10)—C(9)	108.1 (2)	C(1)—C(10)—C(20)	110.3 (2)
C(5)—C(10)—C(20)	111.1 (2)	C(9)—C(10)—C(20)	110.0 (2)
C(9)—C(11)—C(12)	113.5 (2)	C(11)—C(12)—C(13)	113.9 (2)
C(12)—C(13)—C(14)	107.4 (2)	C(12)—C(13)—C(15)	112.7 (2)
C(14)—C(13)—C(15)	106.8 (2)	C(12)—C(13)—C(17)	110.0 (2)
C(14)—C(13)—C(17)	106.4 (2)	C(15)—C(13)—C(17)	113.1 (2)
C(8)—C(14)—C(13)	114.7 (2)	C(13)—C(15)—C(16)	118.3 (2)
C(13)—C(15)—O(2)	108.3 (2)	C(16)—C(15)—O(2)	110.7 (2)
C(15)—C(16)—O(3)	112.1 (2)	C(13)—C(17)—O(4)	115.4 (2)

Table 3. Hydrogen-bonding geometry (\AA , $^\circ$)

$D-H \cdots A$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O(4)—H \cdots O(3)	2.583 (4)	1.68 (3)	164 (2)
O(1)—H \cdots O(2')	2.711 (3)	1.77 (3)	172 (3)
O(3)—H \cdots O(1'')	2.683 (3)	1.86 (3)	161 (3)
O(2)—H \cdots O(4''')	2.636 (3)	1.82 (2)	171 (3)

Symmetry codes: (i) $\frac{3}{2} - x, 1 - y, z - \frac{1}{2}$; (ii) $\frac{3}{2} - x, 1 - y, \frac{1}{2} + z$; (iii) $2 - x, \frac{1}{2} + y, \frac{1}{2} - z$.

The H atoms in the CH and CH₂ groups were allowed to ride on the bonded C atom and refined.

Data collection: *SHELXTL-Plus* (Sheldrick, 1991). Cell refinement: *SHELXTL-Plus*. Data reduction: *SHELXTL-Plus*. Program(s) used to solve structure: *SHELXTL-Plus*. Molecular graphics: *SHELXTL-Plus*. Software used to prepare material for publication: *SHELXTL-Plus*, *PARST* (Nardelli, 1983).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: PT1035). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
 Jiménez, M. & González de la Parra, M. (1983). *Rev. Latinoam. Quím.* **14**, 20–23.
 Larson, A. C. (1967). *Acta Cryst.* **23**, 664–665.
 Nardelli, M. (1983). *Comput. Chem.* **7**, 95–98.
 Rao, Ch. B., Krishna, P. G. & Suseela, K. (1985). *Ind. J. Chem.* **24B**, 403–407.
 Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Soriano-García, M., Guerrero, C. & Toscano, R. A. (1986). *Acta Cryst.* **C42**, 729–731.

Acta Cryst. (1996). **C52**, 1460–1462

Acetonide-3-trinervinone†

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Abstract

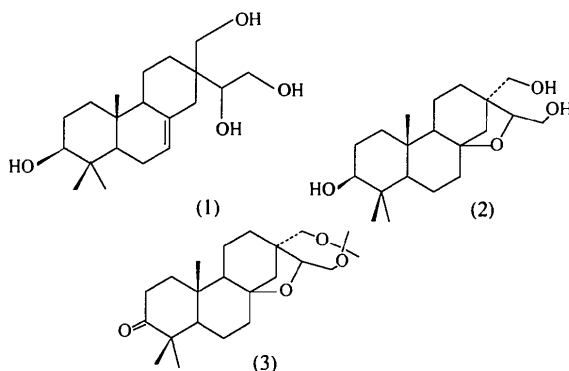
This X-ray diffraction study establishes the molecular structure of the title compound, 2,2,4b',8',8'-penta-methyl-6,10a'-epoxyspiro[1,3-dioxepane-5,2'-perhydro-phenanthren]-7'-one, C₂₃H₃₆O₄, which consists of three

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six-membered rings (*A*, *B* and *C*), and a five- and a seven-membered ring (*D* and *E*, respectively). The *C* ring is fused at C(8)—C(14)—C(13) to the *D* ring. The *A/B*, *B/C* and *D/E* rings are *trans*, *trans* and *cis* fused, respectively. The *A*, *B* and *C* rings adopt a distorted chair 1C_4 conformation. The *D* and *E* rings have envelope and half-chair 5H_4 conformations, respectively. The crystal structure is stabilized by van der Waals interactions. There are two C—H...O hydrogen bonds of less than 3.5 Å.

Comment

Given our interest in the phytochemical study of the Compositae family of plants found in Mexico, we have investigated the constituents of *Piqueria trinervia*, Cav (Jiménez & González de la Parra, 1983). A tetraol tricyclic diterpenoid, trinervinol, (1), was isolated from the flowers and leaves of this plant. During structure elucidation of (1) using chemical transformations and spectral data, compounds (2) and (3) were obtained. Trinervinol is a new diterpene with an isopimarane skeleton with a hydroxyl group at C(17). Details of the chemical transformations and spectral data of (2) and (3) are given by Jiménez & González de la Parra (1983). In this paper, the molecular structure of (3) is reported.



The Cremer & Pople (1975) ring-puckering parameters for the *A*, *B*, *C*, *D* and *E* rings are: *A*, $q^2 = 0.116(5)$, $q^3 = 0.499(5)$, $Q = 0.512(5)$ Å, $\theta = 13.1(5)$, $\varphi = -2(2)^\circ$; *B*, $q^2 = 0.162(4)$, $q^3 = 0.534(4)$, $Q = 0.558(4)$ Å, $\theta = 16.9(4)$, $\varphi = -1(1)^\circ$; *C*, $q^2 = 0.221(4)$, $q^3 = 0.621(5)$, $Q = 0.659(4)$ Å, $\theta = 19.5(4)$, $\varphi = -50.2(9)^\circ$; *D*, $q^2 = 0.445(3)$ Å, $\varphi^2 = 45.2(4)^\circ$; *E*, $q^2 = 0.632(3)$, $q^3 = 0.575(4)$, $Q = 0.854(3)$ Å, $\theta = 47.7(2)$, $\varphi^2 = -35.2(4)$, $\varphi^3 = 167.2(3)^\circ$. A similar conformation has been observed in (3*R*,5*R*,8*S*,10*R*,13*R*)-*ent*-pimara-9(11),15-dien-3-yl *p*-bromobenzoate (Soriano-García, Guerrero & Toscano, 1986) and closely resembles 11β-hydroxy-7-ketosandaracopimar-8(14),15-diene (Rao, Krishna & Suseela, 1985) for the C atoms in rings *A* and *B*. The molecules in the crystal are packed at normal van

der Waals distances. There are two C—H...O interactions of less than 3.5 Å: C(20)...O(2) 3.187(5) and C(23)...O(1)($x - 1, y - 1, z$) 3.463(6) Å.

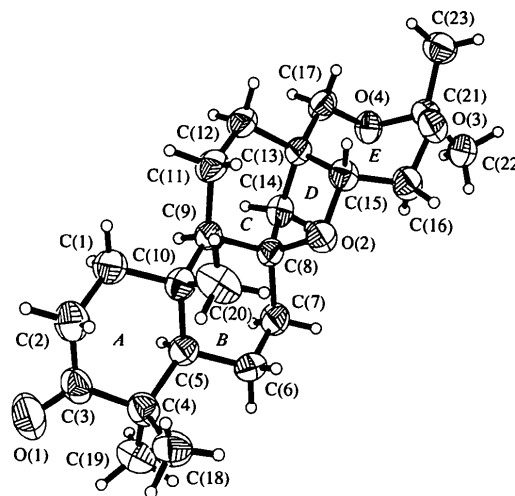


Fig. 1. The molecular structure of the title compound with the atom labelling; 50% probability displacement ellipsoids are shown.

Experimental

Compound (3) was obtained during the elucidation of (1), which was isolated from the flowers and leaves of *Piqueria trinervia*, Cav.

Crystal data

$C_{23}H_{36}O_4$
 $M_r = 376.54$
 Orthorhombic
 $P2_12_1$
 $a = 8.664(2)$ Å
 $b = 12.314(4)$ Å
 $c = 19.763(4)$ Å
 $V = 2108(1)$ Å³
 $Z = 4$
 $D_x = 1.19$ Mg m⁻³
 $D_m = 1.20$ Mg m⁻³
 D_m measured by pycnometry

Mo $K\alpha$ radiation
 $\lambda = 0.71073$ Å
 Cell parameters from 25 reflections
 $\theta = 5.0$ – 20.2°
 $\mu = 0.074$ mm⁻¹
 $T = 296$ K
 Prism
 $0.60 \times 0.52 \times 0.30$ mm
 Colourless

Data collection

Nicolet P3/F diffractometer
 $\theta/2\theta$ scans
 Absorption correction: none
 2297 measured reflections
 2153 independent reflections
 1742 observed reflections
 $[I > 2.8\sigma(I)]$
 $\theta_{max} = 25^\circ$

$h = 0 \rightarrow 10$
 $k = 0 \rightarrow 14$
 $l = 0 \rightarrow 23$
 2 standard reflections monitored every 150 reflections
 frequency: 90 min
 intensity decay: 5%

Refinement

Refinement on F
 $R = 0.052$
 $wR = 0.073$

$(\Delta/\sigma)_{max} = 0.097$
 $\Delta\rho_{max} = 0.223$ e Å⁻³
 $\Delta\rho_{min} = -0.217$ e Å⁻³

S = 1.208

1742 reflections

244 parameters

H-atom parameters not refined

$$w = 1/[\sigma^2(F_o) + 0.00223(F_o)^2]$$

Extinction correction:

Larson (1967)

Extinction coefficient:

0.00062 (3)

Atomic scattering factors

from *SHELXTL-Plus*

(Sheldrick, 1991)

C(11)—C(12)—C(13)	113.2 (3)	C(22)—C(21)—O(3)	113.1 (3)
C(12)—C(13)—C(15)	109.9 (3)	C(22)—C(21)—O(4)	104.8 (3)
C(12)—C(13)—C(17)	107.9 (3)	O(3)—C(21)—O(4)	110.5 (3)
C(15)—C(13)—C(17)	114.9 (3)	C(16)—O(3)—C(21)	115.7 (3)
C(1)—C(2)—C(3)	111.9 (4)	C(13)—C(15)—O(2)	105.9 (3)
C(2)—C(3)—O(1)	120.7 (4)	C(15)—C(16)—O(3)	111.6 (3)
C(3)—C(4)—C(5)	110.2 (3)	C(22)—C(21)—C(23)	110.6 (3)
C(5)—C(4)—C(18)	115.3 (3)	C(23)—C(21)—O(3)	105.8 (3)
C(5)—C(4)—C(19)	108.6 (3)	C(23)—C(21)—O(4)	112.3 (3)
C(4)—C(5)—C(6)	114.0 (3)	C(8)—O(2)—C(15)	109.3 (2)
C(6)—C(5)—C(10)	110.3 (3)	C(17)—O(4)—C(21)	115.3 (3)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$U_{eq} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	x	y	z	U _{eq}
C(1)	1.0563 (4)	0.1689 (4)	0.8590 (3)	0.064 (2)
C(2)	1.1222 (5)	0.2827 (3)	0.8749 (3)	0.071 (2)
C(3)	1.0307 (5)	0.3702 (3)	0.8424 (2)	0.051 (1)
C(4)	0.8553 (5)	0.3693 (3)	0.8533 (2)	0.049 (1)
C(5)	0.7914 (4)	0.2499 (3)	0.8482 (2)	0.042 (1)
C(6)	0.6219 (5)	0.2388 (3)	0.8671 (2)	0.053 (1)
C(7)	0.5574 (4)	0.1333 (3)	0.8389 (2)	0.052 (1)
C(8)	0.6468 (4)	0.0314 (3)	0.8574 (2)	0.040 (1)
C(9)	0.8245 (4)	0.0478 (3)	0.8534 (2)	0.039 (1)
C(10)	0.8884 (4)	0.1570 (3)	0.8818 (2)	0.043 (1)
C(11)	0.9036 (4)	-0.0532 (3)	0.8841 (2)	0.054 (1)
C(12)	0.8316 (4)	-0.1613 (3)	0.8595 (3)	0.052 (1)
C(13)	0.6537 (4)	-0.1597 (3)	0.8592 (2)	0.041 (1)
C(14)	0.6023 (4)	-0.0657 (3)	0.8145 (2)	0.044 (1)
C(15)	0.5932 (4)	-0.1193 (3)	0.9278 (2)	0.044 (1)
C(16)	0.4265 (5)	-0.1489 (3)	0.9429 (2)	0.051 (1)
C(17)	0.5982 (4)	-0.2717 (3)	0.8388 (2)	0.049 (1)
C(18)	0.8282 (6)	0.4260 (3)	0.9217 (2)	0.064 (2)
C(19)	0.7785 (6)	0.4383 (3)	0.7979 (2)	0.063 (2)
C(20)	0.8788 (6)	0.1594 (3)	0.9605 (2)	0.065 (2)
C(21)	0.3541 (4)	-0.3178 (3)	0.8901 (2)	0.045 (1)
C(22)	0.1856 (5)	-0.2966 (4)	0.8750 (3)	0.060 (2)
C(23)	0.3819 (5)	-0.4389 (3)	0.9026 (2)	0.061 (2)
O(1)	1.0927 (4)	0.4387 (3)	0.8087 (2)	0.080 (1)
O(2)	0.6030 (3)	-0.0023 (2)	0.9253 (1)	0.047 (1)
O(3)	0.4066 (3)	-0.2641 (2)	0.9489 (1)	0.051 (1)
O(4)	0.4346 (3)	-0.2790 (2)	0.8318 (1)	0.047 (1)

Table 2. Geometric parameters (Å, °)

C(1)—C(2)	1.546 (6)	C(2)—C(3)	1.484 (6)
C(3)—C(4)	1.535 (6)	C(4)—C(5)	1.574 (5)
C(5)—C(6)	1.522 (5)	C(6)—C(7)	1.521 (6)
C(7)—C(8)	1.520 (5)	C(8)—C(9)	1.554 (5)
C(1)—C(10)	1.530 (5)	C(5)—C(10)	1.568 (5)
C(9)—C(10)	1.559 (5)	C(9)—C(11)	1.545 (5)
C(11)—C(12)	1.548 (5)	C(12)—C(13)	1.541 (5)
C(8)—C(14)	1.515 (5)	C(13)—C(14)	1.522 (5)
C(13)—C(15)	1.536 (5)	C(15)—C(16)	1.519 (6)
C(13)—C(17)	1.515 (5)	C(4)—C(18)	1.540 (6)
C(4)—C(19)	1.538 (6)	C(10)—C(20)	1.558 (5)
C(21)—C(22)	1.512 (5)	C(21)—C(23)	1.531 (5)
C(3)—O(1)	1.202 (5)	C(8)—O(2)	1.456 (4)
C(15)—O(2)	1.443 (4)	C(16)—O(3)	1.434 (4)
C(21)—O(3)	1.413 (5)	C(17)—O(4)	1.426 (4)
C(21)—O(4)	1.429 (5)		
C(2)—C(1)—C(10)	112.2 (3)	C(6)—C(7)—C(8)	115.5 (3)
C(2)—C(3)—C(4)	117.6 (3)	C(7)—C(8)—C(14)	112.8 (3)
C(4)—C(3)—O(1)	121.7 (4)	C(7)—C(8)—O(2)	108.9 (3)
C(3)—C(4)—C(18)	105.7 (3)	C(14)—C(8)—O(2)	103.0 (3)
C(3)—C(4)—C(19)	109.0 (3)	C(8)—C(9)—C(11)	108.3 (3)
C(18)—C(4)—C(19)	108.0 (3)	C(11)—C(9)—C(10)	108.3 (3)
C(4)—C(5)—C(10)	117.7 (3)	C(5)—C(10)—C(9)	106.6 (3)
C(5)—C(6)—C(7)	109.9 (3)	C(5)—C(10)—C(20)	112.4 (3)
C(7)—C(6)—C(9)	112.6 (3)	C(9)—C(10)—C(12)	113.0 (3)
C(9)—C(8)—C(14)	109.1 (3)	C(12)—C(13)—C(14)	107.7 (3)
C(9)—C(8)—O(2)	110.1 (3)	C(14)—C(13)—C(15)	99.5 (3)
C(8)—C(9)—C(10)	116.5 (3)	C(14)—C(13)—C(17)	116.5 (3)
C(10)—C(9)—C(11)	113.3 (3)	C(8)—C(14)—C(13)	101.6 (3)
C(1)—C(10)—C(9)	108.3 (3)	C(13)—C(15)—C(16)	114.8 (3)
C(1)—C(10)—C(20)	110.1 (3)	C(16)—C(15)—O(2)	107.6 (3)
C(9)—C(10)—C(20)	111.0 (3)	C(13)—C(17)—O(4)	113.5 (3)

The H atoms in the CH and CH₂ groups were allowed to ride on the bonded C atom and refined.

Data collection: *SHELXTL-Plus* (Sheldrick, 1991). Cell refinement: *SHELXTL-Plus*. Data reduction: *SHELXTL-Plus*. Program(s) used to solve structure: *SHELXTL-Plus*. Program(s) used to refine structure: *SHELXTL-Plus*. Molecular graphics: *SHELXTL-Plus*. Software used to prepare material for publication: *SHELXTL-Plus*, *PARST* (Nardelli, 1983).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: PT1039). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
 Jiménez, M. & González de la Parra, M. (1983). *Rev. Latinoam. Quím.* **14**, 20–23.
 Larson, A. C. (1967). *Acta Cryst.* **23**, 664–665.
 Nardelli, M. (1983). *Comput. Chem.* **7**, 95–98.
 Rao, Ch. B., Krishna, P. G. & Suseela, K. (1985). *Ind. J. Chem.* **24B**, 403–407.
 Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
 Soriano-García, M., Guerrero, C. & Toscano, R. A. (1986). *Acta Cryst.* **C42**, 729–731.

Acta Cryst. (1996). **C52**, 1462–1464

1-Phenyl-4-imidazolidinone (Z)-Oxime

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Abstract

The treatment of 4-nitroimidazoles with an excess of sodium borohydride in the presence of sodium methoxide yields the (Z)-oximes of 4-imidazolidinones. The title compound, 1-phenyl-4-imidazolidinone (Z)-